## 140a HW 3 revised, Due 1/29/19

 $\star$  All numbered exercises are from Blundell and Blundell.

- 1. Suppose that  $df = xe^{-y}dx + g(x,y)dy$  is an exact differential.
  - (a) What is g(x, y)?

(b) What is  $\int_{\Gamma} df$  where  $\Gamma$  is a semicircle of radius 5 that starts at (x, y) = (0, 0) and ends at (x, y) = (10, 0). Hint: do not actually integrate over  $\Gamma$  but, instead, argue that you can get the same answer by integrating over an easier path, and do that integral.

- $2.\ 12.3.$
- 3. (a) Calculate the work, in J, that is produced when 100g of liquid water vaporizes into steam at  $100^{\circ}C$  against a pressure of one atmosphere (which is the same as the vapor pressure of steam at  $100^{\circ}C$ ). The densities of water and steam at this pressure and temperature are  $0.958g/cm^3$  and  $0.598kg/m^3$ , respectively.

(b) What energy change is involved in the process? The latent heat of vaporation (i.e. the added energy cost to convert from liquid to gas) is 2257J/g.

4. An ideal diatomic gas initially has  $p_i = 4 \times 10^5 Pa$  and  $V_i = 2m^3$  and  $T_i = 293K$ . It undergoes a reversible process with final pressure  $p_f = 4p_i$ .

(a) Suppose that the process is reversible and isothermal. What is  $V_f$ ? Compute  $\Delta U$ ,  $\Delta W$ , and  $\Delta Q$  for the process, in J.

(b) Suppose instead the process is reversible and adiabatic. What is  $V_f$ . Compute  $\Delta U$ ,  $\Delta Q$ , and  $\Delta W$ .

- 5. 13.4.
- 6. 13.5.